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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/530,554	04/07/2005	· Hiroyuki Sato	10936-86	8861
	EXAMINER MESH, GENNADIY			
			ART UNIT	PAPER NUMBER
<b></b>			1711	
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			05/23/2007	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)
	10/530,554	SATO ET AL.
Office Action Summary	Examiner	Art Unit
·	Gennadiy Mesh	1711
The MAILING DATE of this commu Period for Reply	nication appears on the cover sheet w	ith the correspondence address
A SHORTENED STATUTORY PERIOD WHICHEVER IS LONGER, FROM THE - Extensions of time may be available under the provision after SIX (6) MONTHS from the mailing date of this con - If NO period for reply is specified above, the maximum - Failure to reply within the set or extended period for rep Any reply received by the Office later than three months.	MAILING DATE OF THIS COMMUNI ns of 37 CFR 1.136(a). In no event, however, may a nmunication. statutory period will apply and will expire SIX (6) MON Now will, by statute, cause the application to become Al	CATION. reply be timely filed ITHS from the mailing date of this communication. BANDONED (35 U.S.C. § 133).
earned patent term adjustment. See 37 CFR 1.704(b). Status		
1) Responsive to communication(s) fi	iled on <i>07 April 2005</i> .	
2a)⊠ This action is <b>FINAL</b> .	2b)☐ This action is non-final.	
3) Since this application is in conditio	•	ters, prosecution as to the merits is
• • • • • • • • • • • • • • • • • • • •	tice under <i>Ex parte Quayle</i> , 1935 C.E	
Disposition of Claims		
4)⊠ Claim(s) <u>1 and 3-20</u> is/are pending	in the application.	
4a) Of the above claim(s) is/		
5) Claim(s) is/are allowed.		
6)⊠ Claim(s) <u>1 and 3-20</u> is/are rejected	l.	
7) Claim(s) is/are objected to.		
8) Claim(s) are subject to restr	riction and/or election requirement.	
Application Papers		
9) The specification is objected to by t	he Examiner.	
10) The drawing(s) filed on is/ar		by the Examiner.
	ection to the drawing(s) be held in abeya	
	ng the correction is required if the drawing	· ·
11) The oath or declaration is objected	· · · · · · · · · · · · · · · · · · ·	
Priority under 35 U.S.C. § 119		
12) Acknowledgment is made of a clair	n for foreign priority under 35 U.S.C.	§ 119(a)-(d) or (f).
a)⊠ All b)□ Some * c)□ None of:		
	y documents have been received.	
	y documents have been received in A	
	s of the priority documents have been	received in this National Stage
• •	ional Bureau (PCT Rule 17.2(a)).	ivad
* See the attached detailed Office act	ion for a list of the certified copies not	received.
Attachment(s)	_	
1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review		Summary (PTO-413) s)/Mail Date
<ol> <li>Notice of Draftsperson's Patent Drawing Review</li> <li>Information Disclosure Statement(s) (PTO/SB/08</li> </ol>		nformal Patent Application
Paper No(s)/Mail Date	6) Other:	• •
S, Patent and Trademark Office		

#### **DETAILED ACTION**

1. Applicant's Amendment filed on April 30,2007 is acknowledged.

Rejection is maintained as it was set forth in Office Action mailed on Nov 30,2006, but altered due to amendment of Claim 1 made by the Applicant.

### Claim Rejections - 35 USC § 103

The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

2. Claims 1, 3 - 6 and 13 - 20 are rejected under 35 U.S.C. 103(a) as being unpatentable over Shinoda (US 5,412,067) in view of Howelton (US 5,342,918).

Regarding Claim 1 Shinoda discloses preparation ( see abstract) process of polyester with desirable MW(molecular weight) from cyclic esters or their mixtures ( see lines 5 –15,column 1), wherein impurities as water and hydroxycarboxylic acids ( including oligomers) are accurately controlled ( thus proton concentration also controlled) with total amount less than 100 ppm (see abstract, lines 40 – 68,column 2, line 5-7,column 3 and line 50,column 6) in order to produce polyester with desirable MW ( see lines 1-5,column 3).

Shinoda is silent about addition of water to polymerization system. However, addition of water in order to start ring-opening polymerization process is known in the art. For example, Howelton teach addition of water (as polymerization initiator) in ring – opening polymerization (see line 20,column1).

Therefore, it would have been obvious to one of ordinary of skill in the art to use purified cyclic ester in order to obtain polyester with desirable MW per teaching of

Shinoda and add water to polymerization system in order to start or/and increase rate of polymerization as it shown by Howelton.

Subject mater claimed by Applicant in Claims 13 – 16 was discussed above.

Also see Shinoda: lines 5-17, column 1; lines 20 – 25, column 6 and Example 6, wherein copolymer of glycolide and other cyclic monomer is disclosed.

Regarding Claims 17 - 20: Shinoda in view of Howelton discloses substantially same process capable of producing substantially same product as a polyester with same MW. It will be reasonable to believe that other properties as viscosity and Yellowness index will be also substantially same. Burden shits to Applicant provide factual results to the contrary.

3. Claims 7-9 are rejected under 35 U.S.C. 103(a) as being unpatentable over Shinoda (US 5,412,067) in view of Howelton (US 5,342,918) as applied to claims 1-6 and 13 –20 above, and further in view of Early (US 6,437,565).

Regarding Claims 7 – 9: as it shown by Shinoda in view of Howelton that impurities and proton concentration –(see above) are controlled factor in preparation process of the polyester with desirable MW, but silent about regressional correlation between proton concentration and specific physical properties govern by MW of the polymer.

However, regressional analysis is a standard tool, routinely used in the art in order to determine relations between control factors and any functional properties. For example, Early discloses use of regressional analysis ( see Fig.4) in order to determine physical properties of the composition.

Application/Control Number: 10/530,554

Art Unit: 1711

Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to use regressional analysis as taught by Early in order to find optimum amount of proton concentration in obtain polyester with desirable properties by production method disclosed by Shinoda in view of Howelton.

4. Claims 10 –12 are rejected under 35 U.S.C. 103(a) as being unpatentable over Shinoda (US 5,412,067) in view of Howelton (US 5,342,918) as applied to claims 1-6 and 13 –20 above, and further in view of Handbook of Thermoplastic Polymers, Chapter 2, pages 80 –94.

Regarding Claim 11 Shinoda in view of Howelton silent about conducting polymerization in closed volume – particularly inside closed tubes.

However, ring –opening polymerization process can be conduct in closed volume, because process is not required evacuation of byproducts from polymerization system due to nature of this process – no volatile byproducts are generated during polymerization process as it disclosed in Handbook of Thermoplastic Polymers, Chapter 2, pages 90 - 94.

Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to conduct polymerization process disclosed by Shinoda in view of Howelton inside closed tubes (as a simple polymerization reactors), as it claimed by Applicant, due to significant reduction of overall cost of production equipment due to simplicity of this type or reactors (tube).

Regarding Claim 12 Shinoda in view of Howelton silent about conducting polymerization in solid state after initial polymer was produced.

However, solid-state polymerization of polyesters in order to increase MW of the polymer is well known in the art and would be obvious extension of polymerization process as it disclosed in Handbook of Thermoplastic Polymers, Chapter 2, pages 80 – 82.

Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to conduct polymerization process disclosed by Shinoda in view of Howelton with following step of solid-state polymerization as it thought in Handbook of Thermoplastic Polymers in order to increase MW of the final polymer.

## Response to Arguments

- 5. Applicant's arguments filed April 30,2007 have been fully considered but they are not persuasive.
- 5.1. Regarding Applicant's arguments related to Claims 1, 3-6 and 13 20 rejected under 35 U.S.C. 103(a) as being unpatentable over Shinoda (US 5,412,067) in view of Howelton (US 5,342,918):
- i) As it was stated in original rejection- Shinoda disclosed process wherein polymer with desirable characteristics is obtained by controlling impurities that contribute to overall proton concentration thus, first Applicant's argument ( see last paragraph, page 9 of Remarks filed on April 30,2007 further "Remarks") is not persuasive.
- ii). Shinoda as a Primary reference does not need teach all elements of Applicant's Claims 1,3-6and 13-20. Adding water to the polymerization system was taught by Secondary reference as Howelton. Thus, Applicants second argument ( see last paragraph on page 9 of Remarks) is not persuasive.

iii) Regarding amount of volatile matter( see Applicant's arguments on page 11, second paragraph of Remarks) – arguments directed to amount of volatile matter is irrelevant to claimed subject matter of Claims 1 and 3 – 20.

5.2. Regarding Applicant's arguments related to Claims 7-9 rejected under 35 U.S.C. 103(a) as being unpatentable over Shinoda (US 5,412,067) in view of Howelton (US 5,342,918) as applied to claims 1-6 and 13 –20 above, and further in view of Early (US 6,437,565).

As it was discussed in original rejection, Shinoda in view of Howelton teach that impurities and proton concentration (see above) are controlled factor in preparation process of the polyester with desirable MW, but silent about regressional correlation between proton concentration and specific physical properties govern by MW of the polymer. Note, that regressional analysis is a standard tool, routinely used in the art in order to determine relations between control factors and any functional properties. For example, Early discloses use of regressional analysis (see Fig.4) in order to determine physical properties of the composition.

Therefore, it would have been obvious to one of ordinary skill in the art at the time of the invention to use regressional analysis as taught by Early in order to find optimum amount of proton concentration in obtain polyester with desirable properties by production method disclosed by Shinoda in view of Howelton.

5.3. Regarding Applicant's Arguments related to Claims 10-12 rejected under 35U.S.C. 103(a) as being unpatentable over Shinoda (US 5,412,067) in view of Howelton

(US 5,342,918) as applied to claims 1-6 and 13 –20 above, and further in view of Handbook of Thermoplastic Polymers, Chapter 2, pages 80 –94:

- i). Regarding Applicant's arguments ( see page 17 of Remarks) conducting ring-opening polymerization in closed volume HTP teach that ring-opening polymerization process can be conduct in closed volume, because process is not required evacuation of byproducts from polymerization system due to nature of this process no volatile byproducts are generated during polymerization process it is obvious, **based on this teaching**, to conduct polymerization in closed volume.
- ii) Note, when ring-opening polymerization is completed- all cyclic structures are opened than low molecular weight polymers can be subjected to SSP (Solid state polymerization) as it routinely done in the art.

#### Conclusion

Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of

the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Gennadiy Mesh whose telephone number is (571) 272 2901. The examiner can normally be reached on 10 a.m - 6 p.m.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, James Seidleck can be reached on (571) 272 1078. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

> Gennadiy Mesh Examiner Art Unit 1711

6h 05/14/07

James J. Seidisch Supervisory Patent Examine: Technology Center 1700